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1,3-Dipolar Cycloaddition of Dipolar Reagents to Bifunctional Bis Olefins in the Presence of Chloramine-T

V. Padmavathi^a; K. Venugopal Reddy^a; A. Padmaja^a; D. Bhaskar Reddy^a Sri Venkateswara University, Tirupati, India

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1,3-DIPOLAR CYCLOADDITION OF DIPOLAR REAGENTS TO BIFUNCTIONAL BIS OLEFINS IN THE PRESENCE OF CHLORAMINE-T

V. Padmavathi, K. Venugopal Reddy, A. Padmaja, and D. Bhaskar Reddy Sri Venkateswara University, Tirupati, India (Received May 3, 2002; accepted July 27, 2002)

The bifunctional bis pyrazolines and bis isoxazolines have been prepared from 1-aroyl-2-styrylsulfonylethenes and 1-arylsulfonyl-2-styrylsulfonylethenes.

Keywords: 1-aroyl-2-styrylsulfonylethenes; 1-arylsulfonyl-2-styrylsulfonylethenes; bis isoxazoles; bis isoxazolines; bis pyrazoles; bis pyrazoles; bis pyrazolines

INTRODUCTION

One of the general methods to accomplish the synthesis of five-membered heterocycles-pyrazolines and isoxazolines is the 1,3-dipolar cycloaddition of ylide to an alkene involving 3+2 principle. Among the ylides, nitrile imines and nitrile oxides have been used extensively as reactive intermediates in carbon-carbon bond forming reaction processes. They can be generated by the dehydrogenation of araldehyde phenylhydrazones and araldoximes with lead tetraacetate, mercuric acetate, 1-chloro benzotriazole, chloramine-T, among others. However, use of the latter for the in situ generation of dipolar reagents has enthused many organic chemists. In fact we have reported earlier on 1,3-dipolar cycloaddition reaction of chloramine-T catalyzed dipolar reagents with different activated mono olefins.

The successful use of chloramine-T in this front prompted us toward the synthesis of bis pyrazolines and bis isoxazolines by the reaction of bis chalcones and bis sulfones as dipolarophlies with nitrile imines and nitrile oxides as 1,3-dipoles.⁸ To continue our interest in this direction,

The authors are grateful to CSIR, New Delhi for financial assistance.

Address correspondence to V. Padmavathi, Department of Chemistry, Sri Venkateswara University, Tirupati-517 502, India. E-mail: vkpuram2001@yahoo.com

a study on the reaction of bifunctional bis olefinic systems activated by both carbonyl and sulfonyl groups with the latter has been undertaken.

RESULTS AND DISCUSSION

The synthetic scheme involves the reaction of 1-aroyl-2-styrylsul-fonylethenes (I) and 1-arylsulfonyl-2-styrylsulfonylethenes (II) with araldehyde phenylhydrazones/araldoximes in the presence of chloramine-T in MeOH (see Scheme 1). When the reaction was carried out with 1 mmol of I/II and 2 mmol of araldehyde phenylhydrazone in the presence of 2 mmol of chloramine-T, a mixture of products were obtained in 3:1 ratio, which were separated through column chromatography (Table I). They were identified as 1-phenyl-3,5-diaryl-2-pyrazolinyl-1'-phenyl-3'-aryl-4'-aroyl/4'-arylsulfonyl-2'-pyrazolinyl-[4,5']sulfones (III/IV, major) and 1-phenyl-3-aryl-4-aroyl/4-arylsulfo-

SCHEME 1

TABLE I Physical Data of Compounds III-XIV

			ield (% Method				Yield (%) Method
Product	m.p. (°C)	A	В	C	Product	m.p. (°C)	A
IIIa	184–186	68	75	63	IXa	122–123	23
IIIb	176-177	65	76	60	IXb	129 - 130	25
IIIc	169 - 171	60	71	62	\mathbf{IXc}	140 - 141	27
IVa	184 - 185	62	69	65	Xa	136 - 137	23
IVb	172 - 174	64	70	61	Xb	138 - 139	24
IVc	169 - 170	73	67	68	\mathbf{Xc}	132 - 133	21
Va	172 - 173	60	68	64	XIa	216-217	67
Vb	164 - 165	63	67	66	XIb	221-223	65
\mathbf{Vc}	159-161	59	65	68	XIc	204-205	78
VIa	140-141	58	62	62	XIIa	218-219	69
VIb	161-163	62	64	60	XIIb	207-208	58
VIc	157-159	68	62	58	XIIc	214-215	64
VIIa	124 - 125	20	_	_	XIIIa	191-192	69
VIIb	133-134	22	_	_	XIIIb	183-184	72
VIIc	131-132	25	_	_	XIIIc	187-189	73
VIIIa	118-119	27	_	_	XIVa	180-182	70
VIIIb	121-122	24	_	_	XIVb	213-214	61
VIIIc	126-127	18	_	_	XIVc	211–212	63

 $[^]aSatisfactory$ elemental analyses were obtained for representative examples of each series C \pm 0.24, H \pm 0.12, N \pm 0.18.

nyl-5-styrylsulfonyl-2-pyrazoline (VII/VIII, minor) by their ¹H NMR spectra (Table II). Compounds III/IV exhibited four doublets at δ 5.78-5.82 (J = 6.1-6.2 Hz), 6.00-6.01 (J = 6.0-6.3 Hz), 6.49-6.52(J = 6.0-6.2 Hz) and 6.58-6.63 (J = 5.9-6.1) for C_4 -H, C_4 -H, C_5 -H and C_{5′}-H, respectively. However, compounds **VII/VIII** showed a doublet at δ 6.80–6.84 ($J \approx 16.0$ Hz), which accounts for $C_{1'}$ -H, while $C_{2'}$ -H merged with aromatic protons and appeared in the region 6.94–7.91 as a multiplet. Apart from this, two doublets at δ 5.81–5.88 (C₄-H, J=6.0-6.2 Hz) and $6.58-6.63 \text{ (C}_5\text{-H}, J = 5.9-6.2 \text{ Hz})$ also were observed in their spectrum. A similar reaction of I/II with analdoxime also produced a mixture of products (3:1) which were separated by column chromatography. They were found to be 3,5-diaryl-2-isoxazolinyl-3'aryl-4'-aroyl/4'-arylsulfonyl-2'-isoxazolinyl-[4,5']sulfones (V/VI, major) and 3-aryl-4-aroyl/4-arylsulfonyl-5-styrylsulfonyl-2-isoxazoline (IX/X, minor) by their ¹H NMR spectra. Compounds **V/VI** exhibited four doublets at δ 5.80–5.86 (J = 6.0–6.3 Hz), 6.10–6.16 (J = 6.0–6.3 Hz), 6.52-6.56 (J = 6.0-6.3 Hz), 6.62-6.65 (J = 6.0-6.3 Hz) which were accounted for $C_{4'}$ -H, C_4 -H, C_5 -H, and $C_{5'}$ -H. On the other hand, compounds **IX/X** showed a doublet at 6.84–6.88 ($J \approx 16.0 \text{ Hz}$) for $C_{1'}$ -H

TABLE II NMR Spectral Data of Compounds III-XIV

Compd.	$^{1}\mathrm{H}\ \mathrm{NMR}\ (\delta,\mathrm{ppm})$
IIIa	5.81 (d, 1H, $C_{4'}$ -H, $J = 6.1$ Hz), 6.00 (d, 1H, C_{4} -H, $J = 6.3$ Hz), 6.54 (d, 1H, $C_{5'}$ -H, $J = 6.0$ Hz), 6.61 (d, 1H, $C_{5'}$ -H, $J = 6.1$ Hz), 7.18–8.20 (m, 30H, ArH)
IIIb	2.40 (s, 3H, H $^-$ CH ₃), 3.92 (s, 3H, H $^-$ OCH ₃), 5.78 (d, 1H, C ₄ '-H, J = 6.1 Hz), 6.02 (d, 1H, C ₄ -H, J = 6.3 Hz), 6.50 (d, 1H, C ₅ -H, J = 6.1 Hz), 6.63 (d, 1H, C ₅ '-H, J = 5.9 Hz), 7.16 $^-$ 8.21 (m, 28H, ArH)
IVa	5.80 (d, 1H, $C_{4'}$ -H, $J = 6.2$ Hz), 6.05 (d, 1H, C_{4} -H, $J = 6.2$ Hz), 6.56 (d, 1H, C_{5} -H, $J = 6.0$ Hz), 6.63 (d, 1H, $C_{5'}$ -H, $J = 6.1$ Hz), 7.18–8.20 (m, 30H, ArH)
IVb	$\begin{array}{l} 2.45 \text{ (s, 3H, H$\stackrel{\blacksquare}{-}$CH}_3\text{), } 3.90 \text{ (s, 3H, H$\stackrel{\blacksquare}{-}$OCH}_3\text{), } 5.82 \text{ (d, 1H, C}_4\text{'-H, }J=6.1 \text{ Hz),} \\ 6.08 \text{ (d, 1H, C}_4\text{-H, }J=6.0 \text{ Hz), } 6.54 \text{ (d, 1H, C}_5\text{-H, }J=6.2 \text{ Hz), } 6.61 \text{ (d, 1H, C}_5\text{'-H, }J=6.0 \text{ Hz), } 7.168.21 \text{ (m, 28H, ArH)} \end{array}$
Va	5.86 (d, 1H, C_4 '-H, $J=6.0$ Hz), 6.10 (d, 1H, C_4 -H, $J=6.2$ Hz), 6.57 (d, 1H, C_5 -H, $J=6.0$ Hz), 6.62 (d, 1H, C_5 '-H, $J=6.1$ Hz), 7.18–8.20 (m, 20H, ArH)
Vb	$\begin{array}{l} 2.42~(\mathrm{s},3\mathrm{H},\mathrm{H}^{}\mathrm{CH_3}),3.82~(\mathrm{s},3\mathrm{H},\mathrm{H}^{}\mathrm{OCH_3}),5.84~(\mathrm{d},1\mathrm{H},\mathrm{C_{4'}}\text{-H},J=6.3~\mathrm{Hz}),\\ 6.06~(\mathrm{d},1\mathrm{H},\mathrm{C_{4'}}\text{-H},J=6.0~\mathrm{Hz}),6.56~(\mathrm{d},1\mathrm{H},\mathrm{C_{5'}}\text{-H},J=6.2~\mathrm{Hz}),6.60~(\mathrm{d},1\mathrm{H},\mathrm{C_{5'}}\text{-H},J=6.1~\mathrm{Hz}),7.208.22~(\mathrm{m},18\mathrm{H},\mathrm{ArH}) \end{array}$
VIa	5.88 (d, 1H, $C_{4'}$ -H, $J = 6.2$ Hz), 6.14 (d, 1H, C_{4} -H, $J = 6.2$ Hz), 6.59 (d, 1H, C_{5} -H, $J = 6.0$ Hz), 6.65 (d, 1H, $C_{5'}$ -H, $J = 6.1$ Hz), 6.99–8.21 (m, 20H, ArH)
VIb	2.35 (s, 3H, H—CH ₃), 3.84 (s, 3H, H—OCH ₃), 5.79 (d, 1H, C ₄ '-H, J = 6.1 Hz), 6.11 (d, 1H, C ₄ ·H, J = 6.3 Hz), 6.56 (d, 1H, C ₅ ·H, J = 6.2 Hz), 6.63 (d, 1H, C ₅ '-H, J = 6.2 Hz), 7.03–8.14 (m, 18H, ArH)
VIIa	5.84 (d, 1H, $C_{4'}$ -H, $J = 6.2$ Hz), 6.60 (d, 1H, C_{5} -H, $J = 5.9$ Hz), 6.80 (d, 1H, $C_{1'}$ -H, $J = 16.1$ Hz), 6.94–7.79 (m, 21H, ArH and $H_{2'}$)
VIIc	2.33 (s, 3H, H \rightarrow CH ₃), 5.81 (d, 1H, C ₄ -H, J = 6.0 Hz), 6.63 (d, 1H, C ₅ -H, J = 6.2 Hz), 6.84 (d, 1H, C _{1′} -H, J = 16.2 Hz), 7.04 \rightarrow 7.91 (m, 14H, ArH and H _{2′})
VIIIa	5.88 (d, 1H, C_4 -H, $J = 6.2$ Hz), 6.58 (d, 1H, C_5 -H, $J = 6.1$ Hz), 6.82 (d, 1H, C_1 -H, $J = 16.0$ Hz), 6.94–7.79 (m, 21H, ArH and H_2)
VIIIc	2.38 (s, 3H, H \rightarrow CH ₃), 5.86 (d, 1H, C ₄ -H, J = 6.1 Hz), 6.62 (d, 1H, C ₅ -H, J = 6.2 Hz), 6.84 (d, 1H, C _{1′} -H, J = 16.2 Hz), 7.04 \rightarrow 7.91 (m, 14H, ArH and H _{2′})
IXa	5.82 (d, 1H, C_4 -H, $J = 6.2$ Hz), 6.54 (d, 1H, C_5 -H, $J = 6.1$ Hz), 6.74 (d, 1H, $C_{1'}$ -H, $J = 16.1$ Hz), 6.94–7.79 (m, 16H, ArH and $H_{2'}$)
IXc	2.33 (s, 3H, H—CH ₃), 5.94 (d, 1H, C ₄ -H, J = 6.2 Hz), 6.63 (d, 1H, C ₅ -H, J = 6.0 Hz), 6.82 (d, 1H, C _{1′} -H, J = 16.0 Hz), 7.04–7.91 (m, 14H, ArH and H _{2′})
Xa	5.96 (d, 1H, C_4 -H, $J = 6.1$ Hz), 6.72 (d, 1H, C_5 -H, $J = 6.2$ Hz), 6.88 (d, 1H, $C_{1'}$ -H, $J = 16.1$ Hz), 7.04 – 7.91 (m, 16H, ArH and $H_{2'}$)
Xc	2.43 (s, 3H, H—CH ₃), 5.92 (d, 1H, C ₄ -H, J = 6.2 Hz), 6.70 (d, 1H, C ₅ -H, J = 6.2 Hz), 6.85 (d, 1H, C _{1′} -H, J = 16.0 Hz), 7.04–7.91 (m, 14H, ArH and H _{2′})

while $C_{2'}$ -H merged with aromatic protons and appeared in the region 6.98–7.80 as a multiplet apart from signals due to isoxazoline ring protons (Table II). The olefinic protons in **VII–X** could exist as cis and trans isomers, but their J values at \sim 16.0 Hz indicate that they possess

only the trans geometry. In III-VI there is a possibility of a mixture of diastereomeric bis adducts. However, we could isolate only one pure compound. A small amount of the other isomers, if any, formed could not be isolated. The IR spectra of III-X displayed bands in the regions 1133–1156 and 1311–1332 (SO₂), and 1444–1456 (C=N). Apart from these, III, V, VII, and IX also exhibited a strong band at 1678–1686 (C=O) and another band around 1630–1644 (C=C) in VII-X. However, repetition of these reactions with I/II (1 mmol), araldehyde phenylhydrazone/araldoxime (2 mmol) and a two-fold excess of chloramine-T resulted only III-VI. On the other hand, when VII-X were treated with equimolar proportions of araldehyde phenylhydrazone/araldoxime and chloramine-T furnished once again **III-VI**. The authenticity of **III-VI**, obtained by different routes, was confirmed by their mmp and ¹H NMR spectra and their yields in each method were given separately in Table I. The compounds **III-VI** on oxidation with chloranil in xylene¹⁰ gave their respective pyrazoles/isoxazoles XI-XIV. The absence of two doublets corresponding to pyrazoline/isoxazoline ring protons in their ¹H NMR spectra confirm the formation of **XI-XIV**.

EXPERIMENTAL

Melting points were determined on a Mel-Temp apparatus and are uncorrected. The purity of the compounds was checked by TLC (silica gel H, BDH, ethyl acetate/hexane, 1:3). The IR spectra were recorded on a Perkin-Elmer grating infrared spectrophotometer ($\nu_{\rm max}$ in cm $^{-1}$) model 337 in KBr pellets. The $^1{\rm H}$ NMR spectra were recorded in CDCl₃/DMSO-d₆ on a Varian EM-360 spectrometer (300 MHz) with TMS as an internal standard. The elemental analyses were performed at Dr. Reddy's Research Foundation, Hyderabad, India. The 1-aroyl-2-styrylsulfonylethenes and 1-arylsulfonyl-2-styrylsulfonylethenes were prepared as per the literature procedure. 11

General Procedure for the Preparation of 1-Phenyl-3,5-diaryl-2-pyrazolinyl-1'-phenyl-3'-aryl-4'-aroyl/4'-arylsulfonyl-2'-pyrazolinyl-[4,5']sulfones-III/IV/1-phenyl-3-aryl-4-aroyl/4-arylsulfonyl-5-styrylsulfonyl-2-pyrazolineVII/VIII

Method A

To a mixture of **I/II** (10 mmol), araldehyde phenylhydrazone (20 mmol) and chloramine-T (20 mmol) in MeOH (20 ml) was added and refluxed for 8–9 h. Inorganic salts were filtered off. The filtrate was

concentrated and the residue was extracted with methylene chloride (30 ml). The organic extract was washed with water (2 \times 25 ml), brine (2 \times 25 ml), and dried (an. Na₂SO₄). Evaporation of the solvent under reduced pressure gave a mixture of products, which were separated through a column of silica gel (BDH, 100–200 mesh) using hexane: ethyl acetate (10:1) as the eluent. They were identified as **III/IV** (major) and **VII/VIII** (minor).

General Procedure for the Preparation of 3,5-Diaryl-2-isoxazolinyl-3'-aryl-4'-aroyl/4'-arylsulfonyl-2'-isoxazolinyl-[4,5']sulfonesV/VI/3-aryl-4-aroyl/4-arylsulfonyl-5-styrylsulfonyl-2-isoxazoline IX/X

Method A

A mixture of **I/II** (10 mmol) in MeOH (20 ml), araldoxime (20 mmol) and chloramine-T (20 mmol) were added and refluxed for 6–7 h. The salts formed, if any were removed by filtration. The filtrate was concentrated and extracted with methylene chloride. The organic extract was washed with water (2×25 ml) and brine (2×25 ml) and dried (an. Na₂SO₄). The solvent was removed in *vacuo*. The mixtures of products obtained were separated by column chromatography using silica gel (BDH, 100–200 mesh) with hexane: ethyl acetate (10:1) as eluent. The compounds were identified as **V/VI** (major) and **IX/X** (minor).

General Procedure for the Preparation of III-VI

Method B

The compound **I/II** (10 mmol), araldehyde phenylhydrazone (20 mmol)/araldoxime (20 mmol) and a two-fold excess of chloramine-T in MeOH (25 ml) was refluxed for 6–7 h. The salts formed were filtered off. The filtrate was concentrated and the residue was extracted with methylene chloride. The organic extract was washed with water (2 \times 25 ml), brine (2 \times 25 ml), and dried (an. Na₂SO₄). Evaporation of the solvent under reduced pressure gave crude product, which was purified through a column of silica gel (BDH, 100–200 mesh) using hexane: ethyl acetate (10:1) as the eluent to provide pure **III–VI**.

Method C

An equimolar (10 mmol) mixture of **VII/VIII/IX/X**, araldehyde phenylhydrazone/araldoxime and chloramine-T in MeOH (20 ml) was refluxed for 5–6 h. The reaction mixture was processed as mentioned previously to provide pure **III–VI**.

Dehydrogenation of III-VI

A mixture of **III/IV/V/VI** (10 mmol) and chloranil (11 mmol) in xylene (20 ml) was refluxed for 30–32 h and the organic layer was separated and washed with 1N NaOH, water and dried (an. Na₂SO₄). The solvent was removed under reduced pressure. The solid obtained was purified by recrystallization from MeOH to provide **XI-XIV** respectively.

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